Altitude

Pressure, Density, Above Sea Level. . .



Why is it so hard to measure?

Know your altitude = don't hit mountains (important)!

How can we measure it at all?

Let's start with a simple hot air balloon:

We begin by hanging a large yardstick off the edge of our balloon as shown. Using this technique.

We touch our yardstick to the ground and measure our distance from the grass/water below us.

Knowing our altitude <u>Above Ground Level (AGL)</u> is great, but we cannot fly around with a large stick hanging from our aircraft!





Understanding Pressure

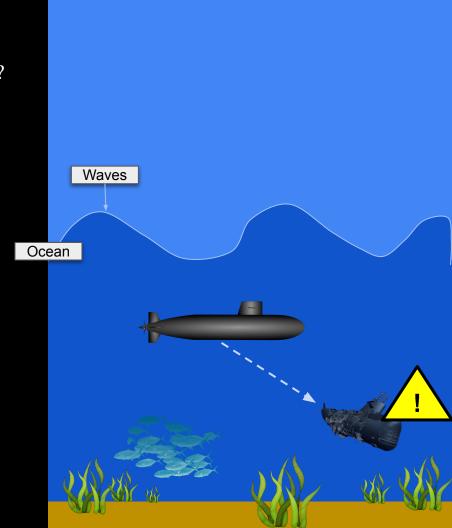
Consider the submarine: Why can it not descend too deep? "Of course", you say, "it will be crushed!"

That's true, the submarine *will* get crushed. But why?

Because of *PRESSURE*.

The WEIGHT of the water above it will smush the submarine, and it will implode!

This "weight" that the submarine would feel, we call <u>Pressure</u>

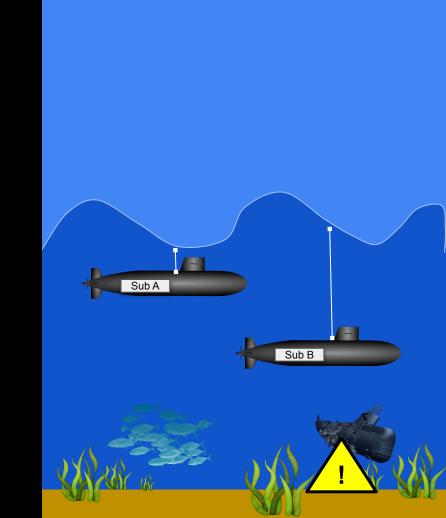


Understanding Pressure

Compare the two submarines.

Submarine A has a small distance below the water, therefore less water "weight" (we know this is called pressure), crushing it.

Comparatively, Submarine B has a large distance below the water, and therefore MORE pressure trying to crush it!



Pressure, the superior way to measure altitude

The Atmosphere is <u>IUST LIKE THE</u> OCEAN!

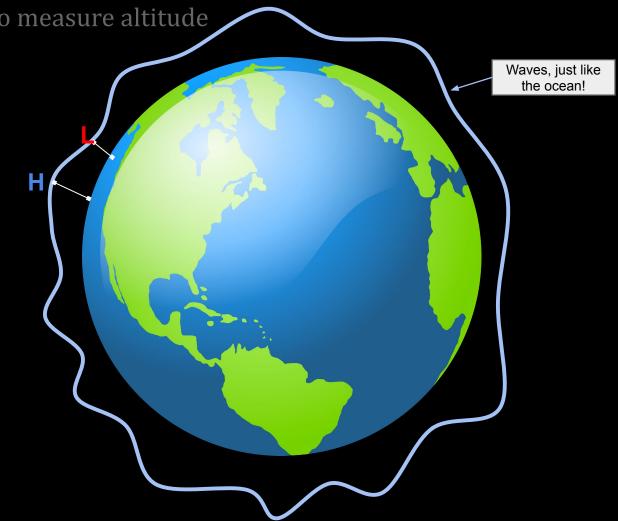
Look at how we can measure the pressure, or "weight" of the AIR just like we did with the submarines.

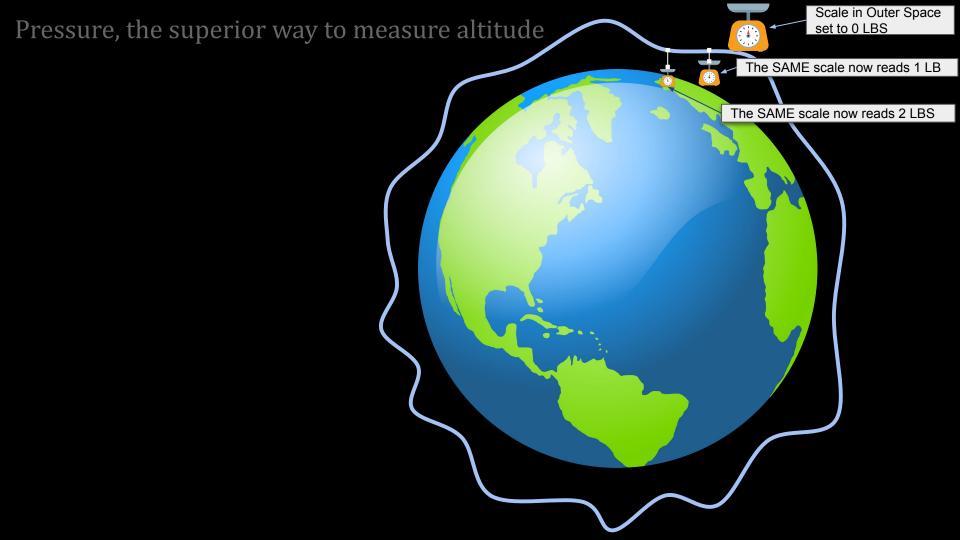
H - is used to signify an area of more weight to "push down" on us.

- is used to signify an area of less weight to "push down" on us.

We call "H" a <u>H</u>igh Pressure Area
We call "L" a <u>L</u>ow Pressure Area

Note: this High and Low pressure theory is very important in meteorology!!

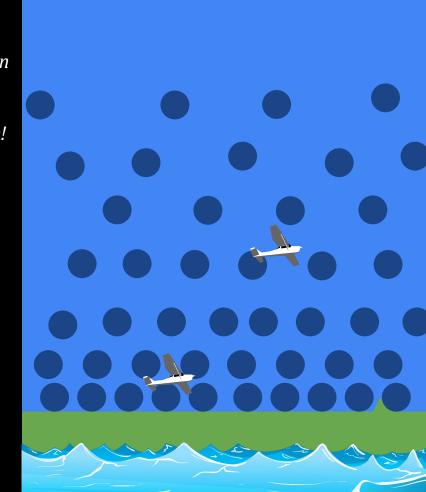




Pressure, the superior way to measure altitude

The Ocean and the Sky share this behavior of "pressure". Now, we *won't* crush our airplane by flying too low, but we *can* measure the change in "weight" as we climb and descend!

This Change in "air weight", or <u>pressure</u> can indicate our altitude!



Now let's look at the *PRESSURE*, or, *WEIGHT* of air!





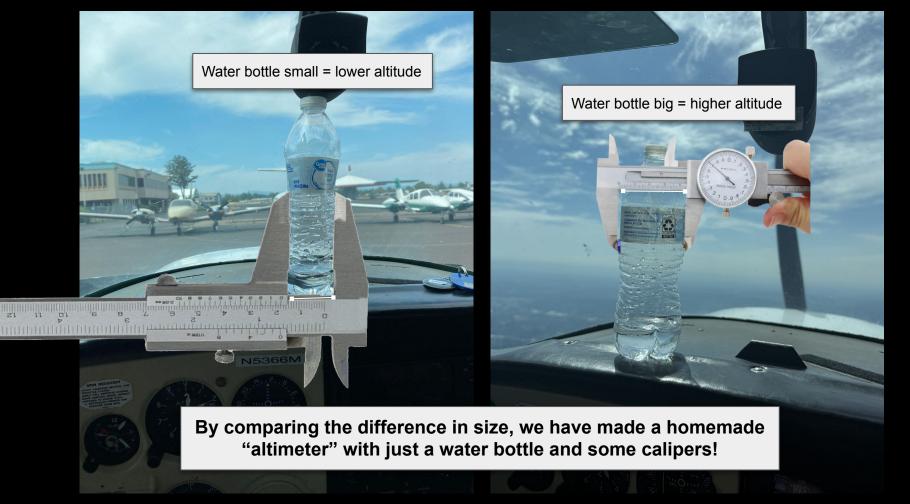


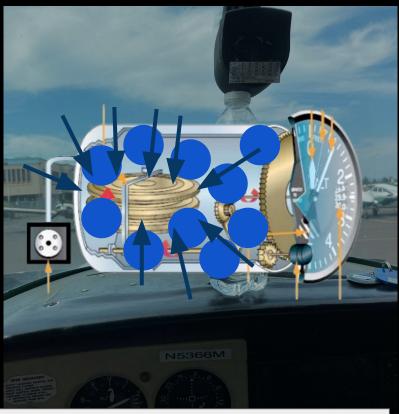
At low altitude, lots of air molecules squash this water bottle, as seen.



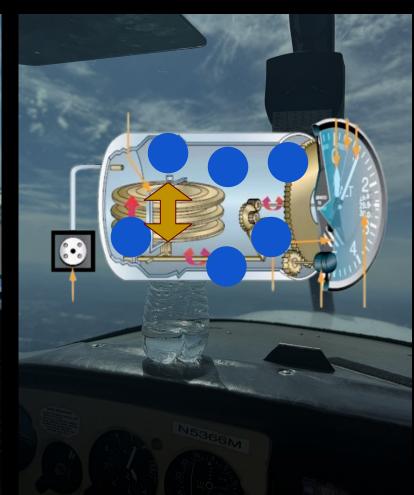
At high altitude, there's not so many air molecules to squash the water bottle, so it is free to expand, and it does!!

Now let's look at the *PRESSURE*, or, *WEIGHT* of air!





At low altitude, lots of air molecules <u>squash this altimeters'</u> <u>diaphragm</u>, as seen. This causes it to shrink in size, and through mechanical linkages: <u>indicate a lower altitude</u>



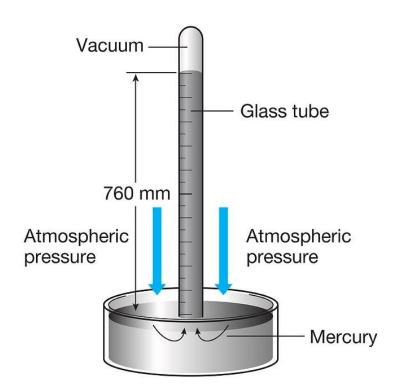
Barometer

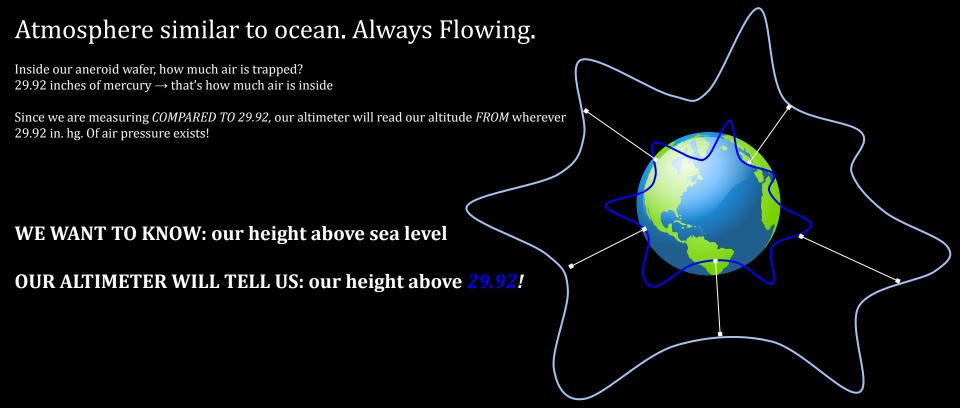
Greek "baros" = weight Greek "metron" = measure

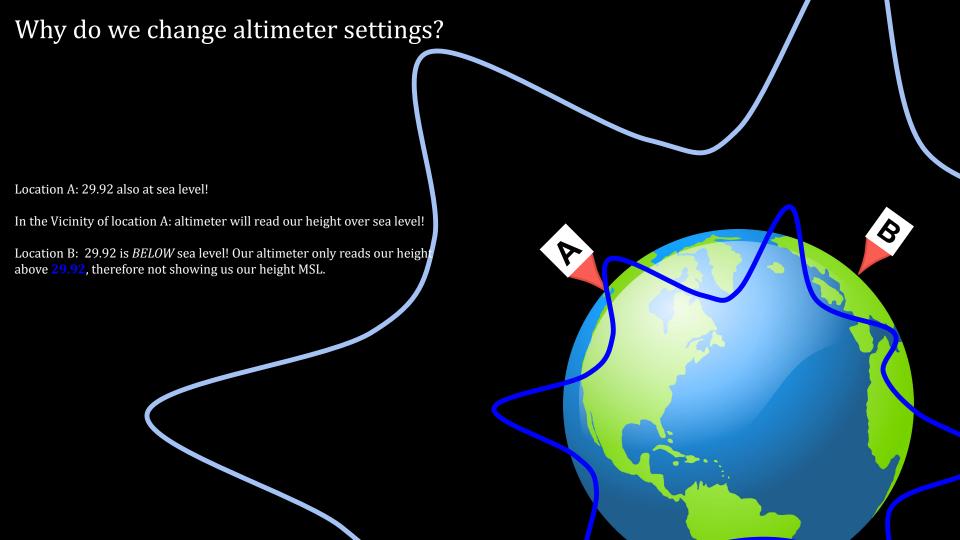
Weight-measurer

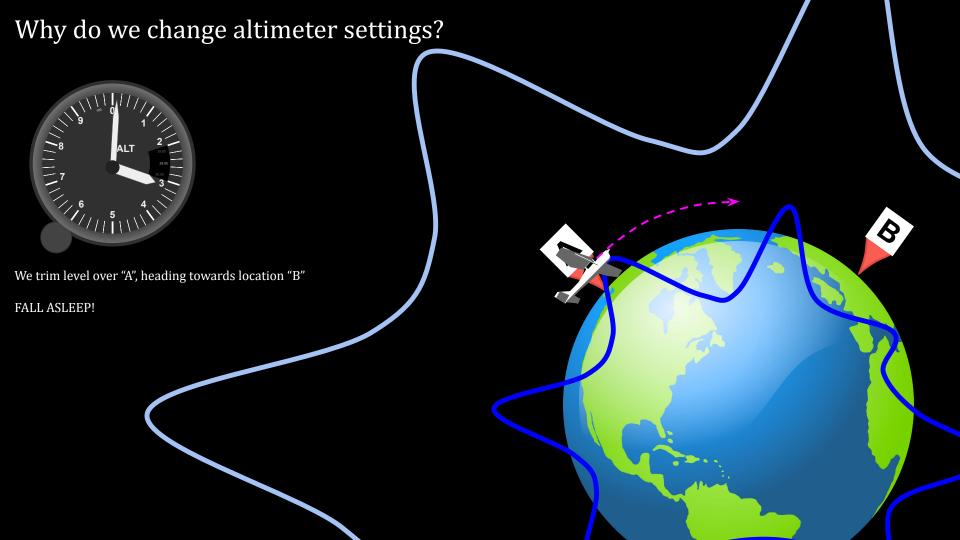
Weight of the air-instrument!

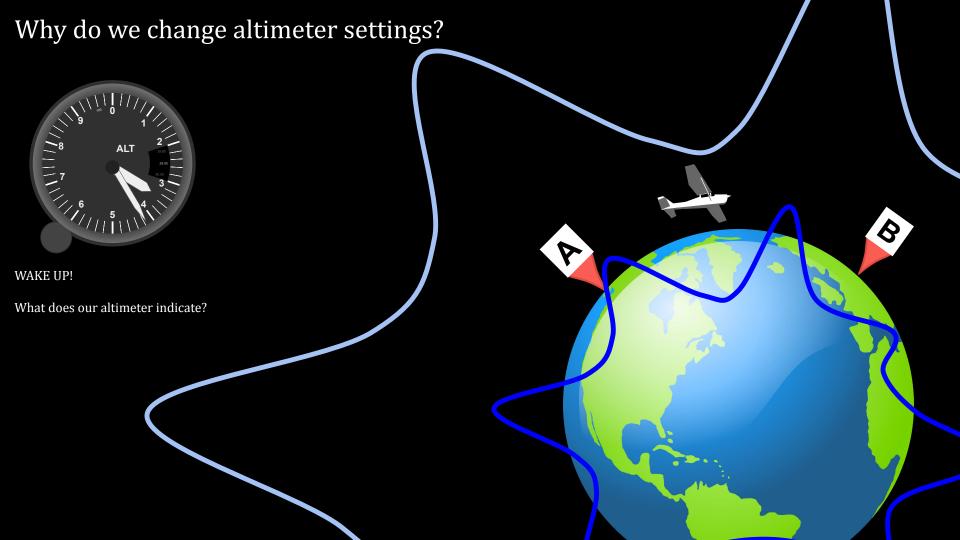
THIS IS USED TO MEASURE AIR PRESSURE

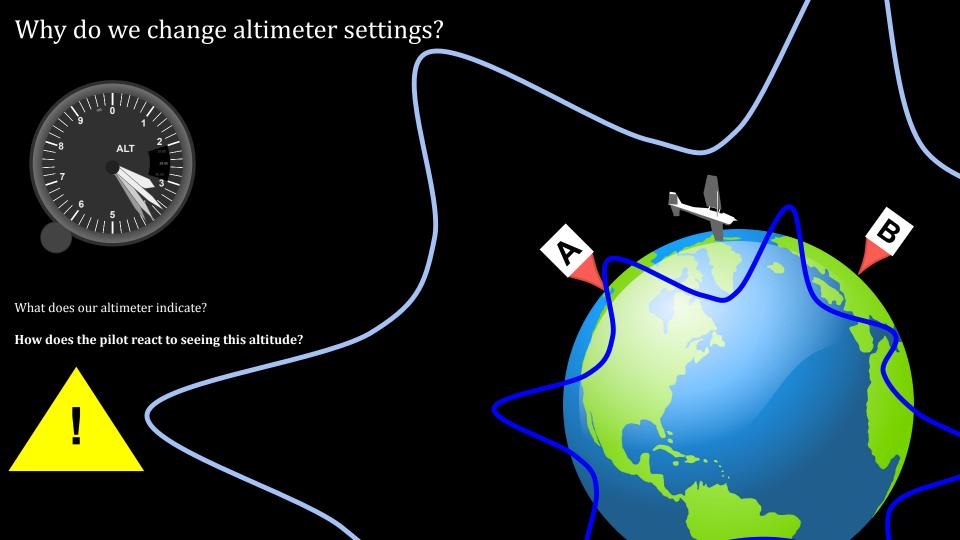


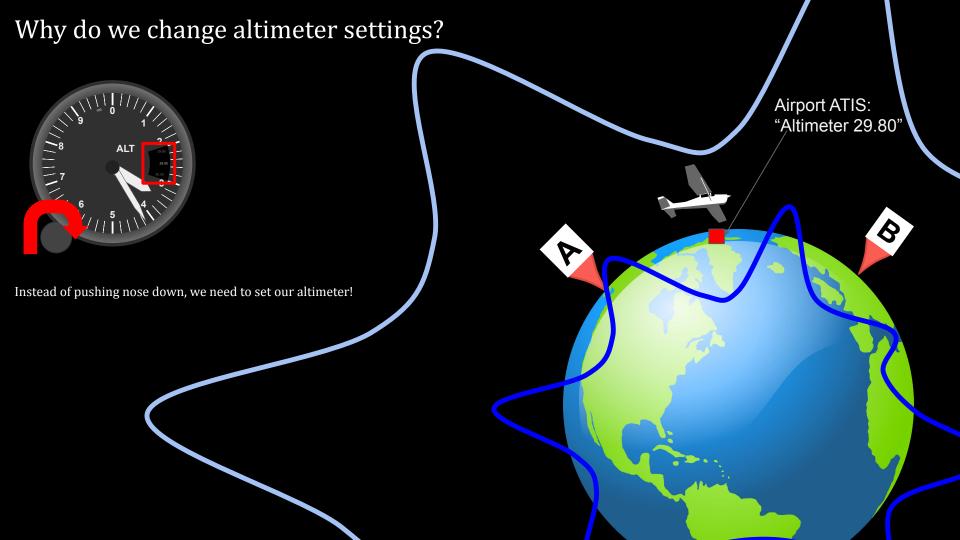


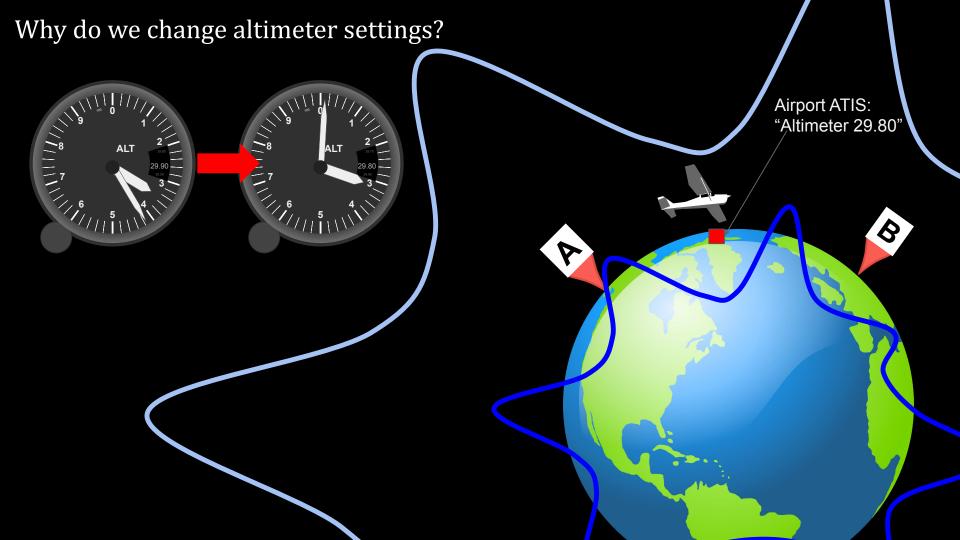


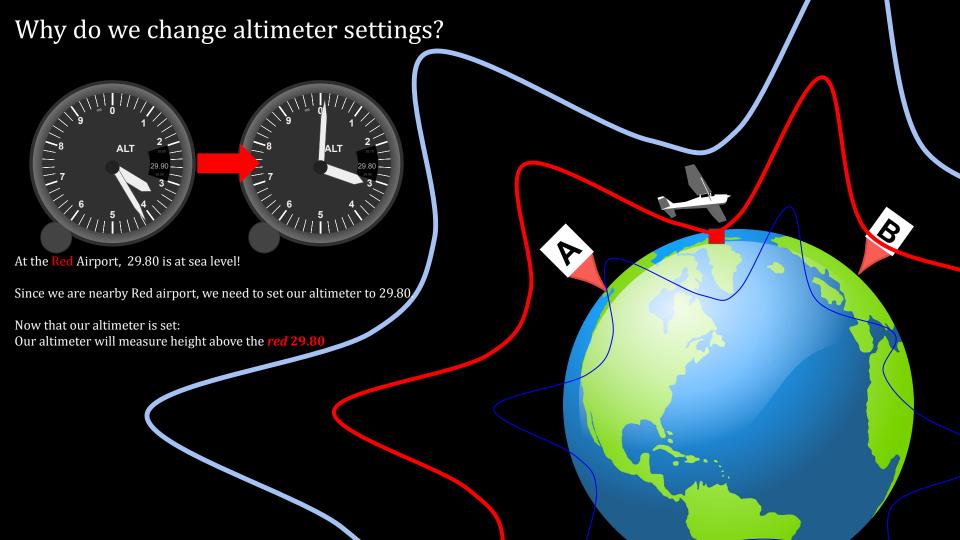


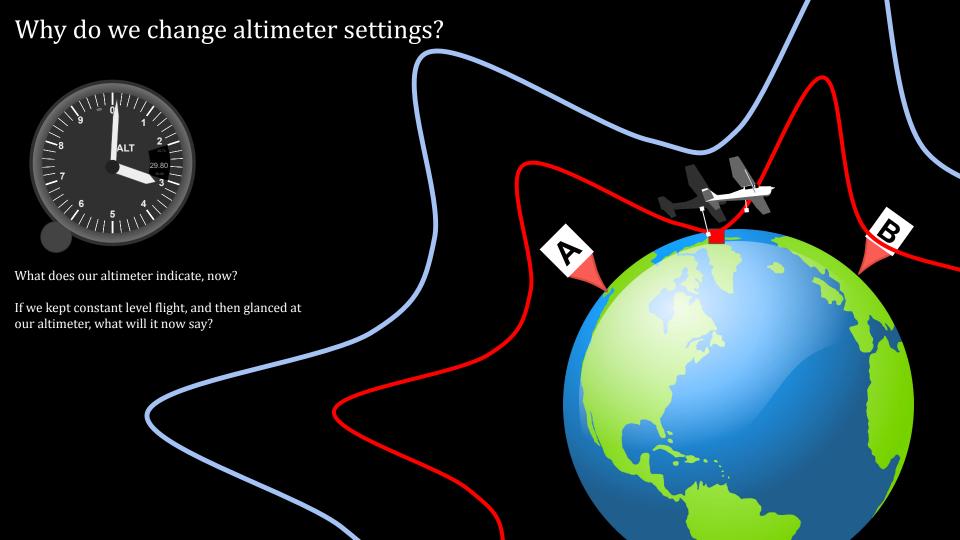


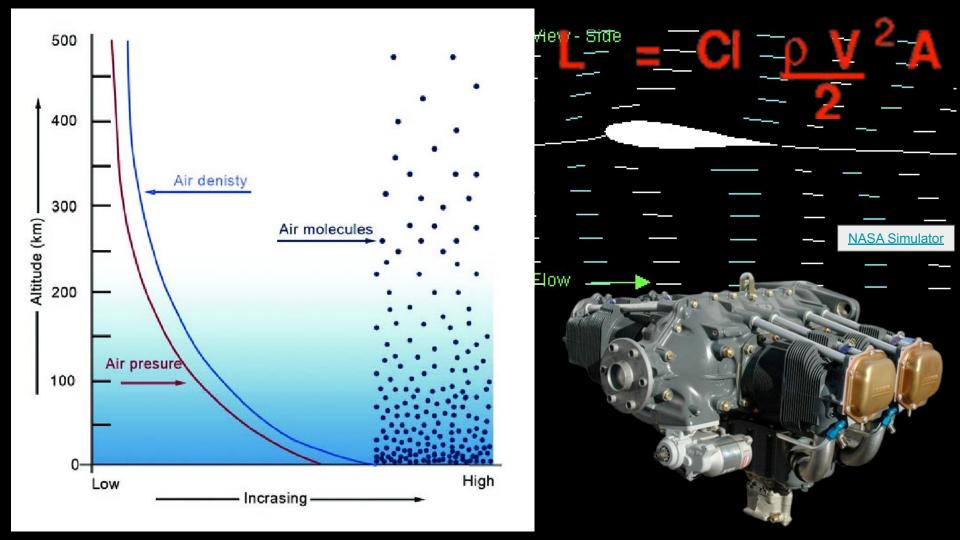


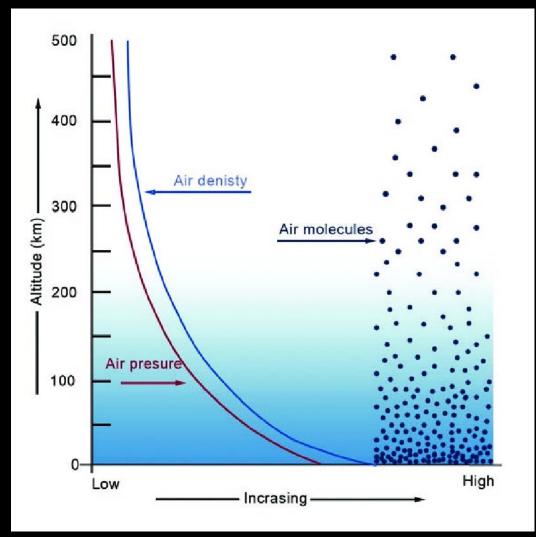










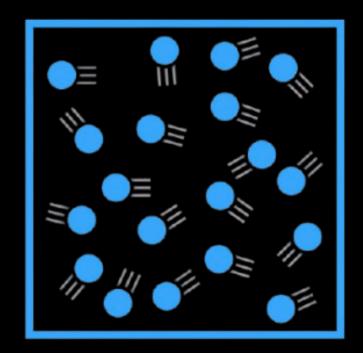


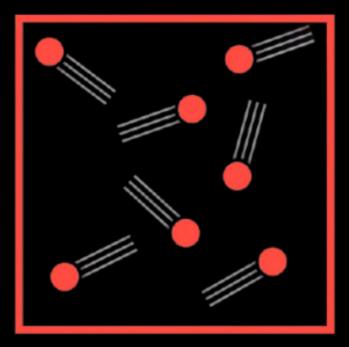


- 1. Oxygen
- 2. Fuel
- 3. Heat

What changes air density?

- 1. Air Pressure
- 2. Temperature
- 3. Humidity



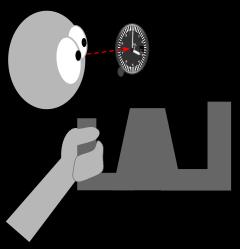


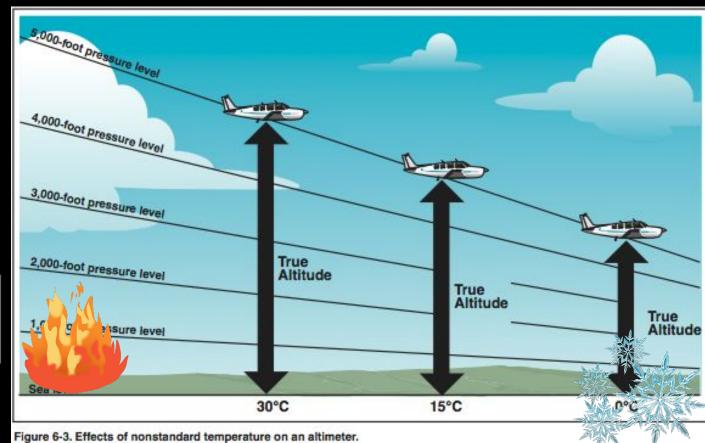
Cold Air

Hot Air

What changes air density?

- 1. Air Pressure
- 2. Temperature
- 3. Humidity

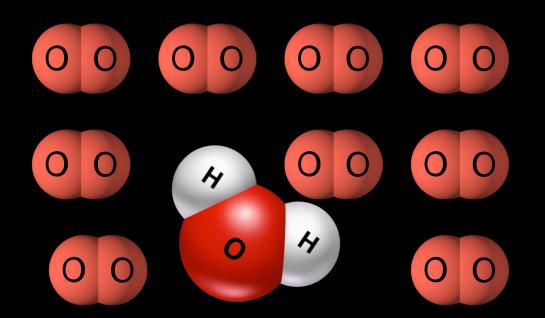


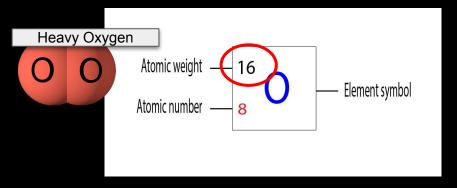


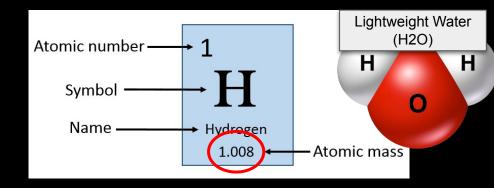
What changes air density?

- 1. Air Pressure
- 2. Temperature
- 3. Humidity*

*Humidity is dependant upon temp. So, it is reasonably accurate to ignore in our everyday calculations.







How *dense* is the air?

We use a formula to relate air density to the altitude the airplane would "feel" like it's at.

Step 1: calculate Pressure altitude (height above standard 29.92)

P.A. = Indicated Altitude + ([29.92 - alt. setting)x1,000)

Step 2: calculate Density Altitude (account for temperature)

D.A. = P.A. + (120x[OAT - standard temp])

(step 2 may be done with E6B)